# organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

# Nam Sook Cho,<sup>a</sup> Soon II Hong,<sup>a</sup> Jin-Gyu Kim<sup>b</sup> and II-Hwan Suh<sup>b</sup>\*

<sup>a</sup>Department of Chemistry, Chungnam National University, Taejon 305-764, Korea, and <sup>b</sup>Department of Physics, Chungnam National University, Taejon 305-764, Korea

Correspondence e-mail: ihsuh@cnu.ac.kr

#### **Key indicators**

Single-crystal X-ray study T = 293 K Mean  $\sigma$ (C–C) = 0.004 Å Disorder in main residue R factor = 0.035 wR factor = 0.089 Data-to-parameter ratio = 12.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 1,5-[5,5'-(3-Oxa-1,5-pentylenedithio)bis(2,3-dihydro-2-oxo-1,3,4-thiadiazol-3-yl)]-3-oxapentane

The title compound, 4,14-dioxa-9,11,17,19-tetrathia-1,7,21,22-tetraazatricyclo[16.2.1.1<sup>7,10</sup>]docosa-18(21),10(22)-diene-8,20-dione,  $C_{12}H_{16}N_4O_4S_4$ , has crystallographic twofold symmetry, with a half molecule in the asymmetric unit and a dihedral angle of 52.59 (6)° between the two planar five-membered rings of the macrocycle, giving the molecule a twisted conformation.

## Comment

As a continuation of our studies of macrocyclic compounds containing 1,3,4-thiadiazole subunits (Cho, Park & Hwang, 1999; Cho, Park, Kim *et al.*, 1999; Cho, Park, Hwang *et al.*, 1999; Cho *et al.*, 2000), we have turned our attention to macrocycles composed of two 5-mercapto-2,3-dihydro-1,3,4-thiadizole-2-one units. For the formation of macrocycle (I), these two heterocyclic rings are linked by ether chains. These compounds have been shown to be artificial receptors of transition metals (Cho, Park, Kim *et al.*, 1999) and furthermore are potential hosts of small organic molecules. The determination of the structure of title compound, (I), is part of a study of these macrocycles.



The title compound, (I), with the atom-labelling scheme is shown in Fig. 1. A half molecule belongs to an asymmetric unit and the other half molecule is generated by applying a twofold rotation [symmetry code: (i) -x, y,  $-z + \frac{3}{2}$ ] passing through O1 and O3. The S–C distances range from 1.674 (5) to 1.970 (9) Å, with a mean value of 1.782 (2) Å, and the N2–C4 and O2–C3 distances of 1.279 (3) and 1.215 (3) clearly show double-bond character. The remainder of the bonds are single bonds; in the five-membered ring, C3–S1–C4 = 88.89 (11)°. These values are comparable with those found in 4,17,25,26-tetraaza-6,9,12,15-tetraoxa-2,19,21,24-tetrathiatricyclo[18.4.1<sup>1,4</sup>.1<sup>17,20</sup>]hexacosa-1(25),20(26)-diene-3,5,16,18-

© 2001 International Union of Crystallography Printed in Great Britain – all rights reserved Received 27 March 2001 Accepted 11 April 2001

Online 26 April 2001



Figure 1

*ORTEP-3* (Farrugia, 1997) diagram of (I) showing 25% probability displacement ellipsoids. C atoms are shown with ellipsoids with only enveloping ellipses and the remaining atoms are shown with ellipsoids with octant shading. Only the asymmetric unit is labelled and H atoms have been omitted for clarity [symmetry code: (i) -x, y,  $-z + \frac{3}{2}$ ].

tetraone (Cho *et al.*, 2000) and 1,5-[5,5'-(1,3-phenylenedimethylenedithio)bis(2,3-dihydro-2-oxo-1,3,4-thiadiazol-3-yl)]-3-oxapentane (Cho *et al.*, 2001).

The five-membered ring, 5-mercapto-2,3-dihydro-1,3,4thiadizole-2-one, is planar within 0.016 (1) Å and the ethyl ether linkage (atoms C2, C1, O1, C1<sup>i</sup> and C2<sup>i</sup>) is planar within 0.085 (2) Å. The dihedral angle between the five-membered ring and that related by twofold symmetry is 52.59 (6)°; the dihedral angle between the five-membered ring and the ethyl ether linkage is 79.02 (6)°. Therefore, the molecule has a twisted conformation with a C2–C1–O1–C1<sup>i</sup> torsion angle of 171.4 (2)°. The interatomic distances N1···N1<sup>i</sup> = 5.609 (4) Å, N2···N2<sup>i</sup> = 4.842 (4) Å, N2···O3 = 4.223 (3) Å, N2···S2<sup>i</sup> = 6.158 (2) Å, O1···O3 = 5.431 (3) Å and S2···S2<sup>i</sup> = 7.322 (2) Å appear to be sufficient to introduce a guest ion inside the molecule. The closest intermolecular distance is O2···H5A = 2.572 (5) Å.

# **Experimental**

The preparation of (I) followed the general procedure previously reported by Cho, Park, Kim *et al.* (1999). Details will be reported elsewhere.

## Crystal data

 $\begin{array}{l} C_{12}H_{16}N_4O_4S_4\\ M_r = 408.53\\ \text{Monoclinic, } C2/c\\ a = 8.7369 (8) \text{ Å}\\ b = 15.741 (3) \text{ Å}\\ c = 13.3967 (12) \text{ Å}\\ \beta = 103.833 (8)^{\circ}\\ V = 1788.9 (4) \text{ Å}^3\\ Z = 4 \end{array}$ 

## Data collection

Enraf-Nonius CAD-4 diffractometer Non-profiled  $\omega/2\theta$  scans Absorption correction:  $\psi$  scan (North *et al.*, 1968)  $T_{min} = 0.745, T_{max} = 0.861$ 3139 measured reflections 1573 independent reflections 1201 reflections with  $I > 2\sigma(I)$  
$$\begin{split} D_x &= 1.517 \text{ Mg m}^{-3} \\ \text{Mo } K\alpha \text{ radiation} \\ \text{Cell parameters from 25} \\ \text{reflections} \\ \theta &= 11.4 - 13.7^{\circ} \\ \mu &= 0.56 \text{ mm}^{-1} \\ T &= 293 \text{ (2) K} \\ \text{Block, colorless} \\ 0.53 \times 0.43 \times 0.26 \text{ mm} \end{split}$$

 $\begin{aligned} R_{\text{int}} &= 0.026\\ \theta_{\text{max}} &= 25.0^{\circ}\\ h &= -10 \rightarrow 10\\ k &= -18 \rightarrow 18\\ l &= -15 \rightarrow 0\\ 3 \text{ standard reflections}\\ \text{frequency: 300 min}\\ \text{intensity decay: none} \end{aligned}$ 

#### Refinement

1

| Refinement on $F^2$             |  |
|---------------------------------|--|
| $R[F^2 > 2\sigma(F^2)] = 0.035$ |  |
| $vR(F^2) = 0.089$               |  |
| S = 1.02                        |  |
| 573 reflections                 |  |
| 30 parameters                   |  |
| I-atom parameters constrained   |  |
|                                 |  |

$$\begin{split} &w = 1/[\sigma^2(F_o^{-2}) + (0.0409P)^2 \\ &+ 0.7080P] \\ &where \ P = (F_o^2 + 2F_c^{-2})/3 \\ (\Delta/\sigma)_{max} < 0.001 \\ \Delta\rho_{max} = 0.16 \ e^{\Lambda^{-3}} \\ \Delta\rho_{min} = -0.20 \ e^{\Lambda^{-3}} \\ &Extinction \ correction: \ SHELXL97 \\ &Extinction \ coefficient: \ 0.0028 \ (6) \end{split}$$

 Table 1

 Selected geometric parameters (Å, °).

| S1-C4    | 1.737 (2)  | S2-C6′ | 1.970 (9) |
|----------|------------|--------|-----------|
| S1-C3    | 1.784 (3)  | N2-C4  | 1.279 (3) |
| S2-C5    | 1.674 (5)  | O2-C3  | 1.215 (3) |
| S2-C4    | 1.747 (3)  |        |           |
| C4-S1-C3 | 88.89 (11) |        |           |

Atoms C5 and C6 were disordered over two positions and their coordinates together with site-occupation factors were refined anisotropically using the *PART* command of *SHELXL*97 (Sheldrick, 1997). All H atoms were located geometrically with  $U_{iso}$  constrained to be 1.2 times the equivalent isotropic displacement of the parent atoms. The highest peak and deepest holes in the final difference density map are at 0.85 Å from H6A and 0.91 Å from S1, respectively.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

## References

- Cho, N. S., Hong, S. I., Choo, G. H., Kim, J.-G. & Suh, I.-H. (2001). Acta Cryst. E57, 0368–369.
- Cho, N. S., Hong, S. I., Kim, J.-K. & Suh, I.-H. (2000). Acta Cryst. C56, 229–230.
  Cho, N. S., Park, H. S. & Hwang, H. J. (1999). Bull. Korean Chem. Soc. 20, 611–613
- Cho, N. S., Park, C. K., Hwang, H. J., Hong, S. I., Park, J. K. & Suh, I.-H. (1999). *J. Chem. Res.* (S), pp. 730–731.
- Cho, N. S., Park, C. K., Kim, H.-S., Oh, J. G., Suh, I.-H. & Oh, M. R. (1999). *Heterocycles*, **51**, 2739–2746.
- Enraf-Nonius (1994). CAD-4 EXPRESS. Enraf-Nonius, Delft, The Netherlands.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). WinGX. University of Glasgow, Scotland.
- Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany. North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351– 359.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.